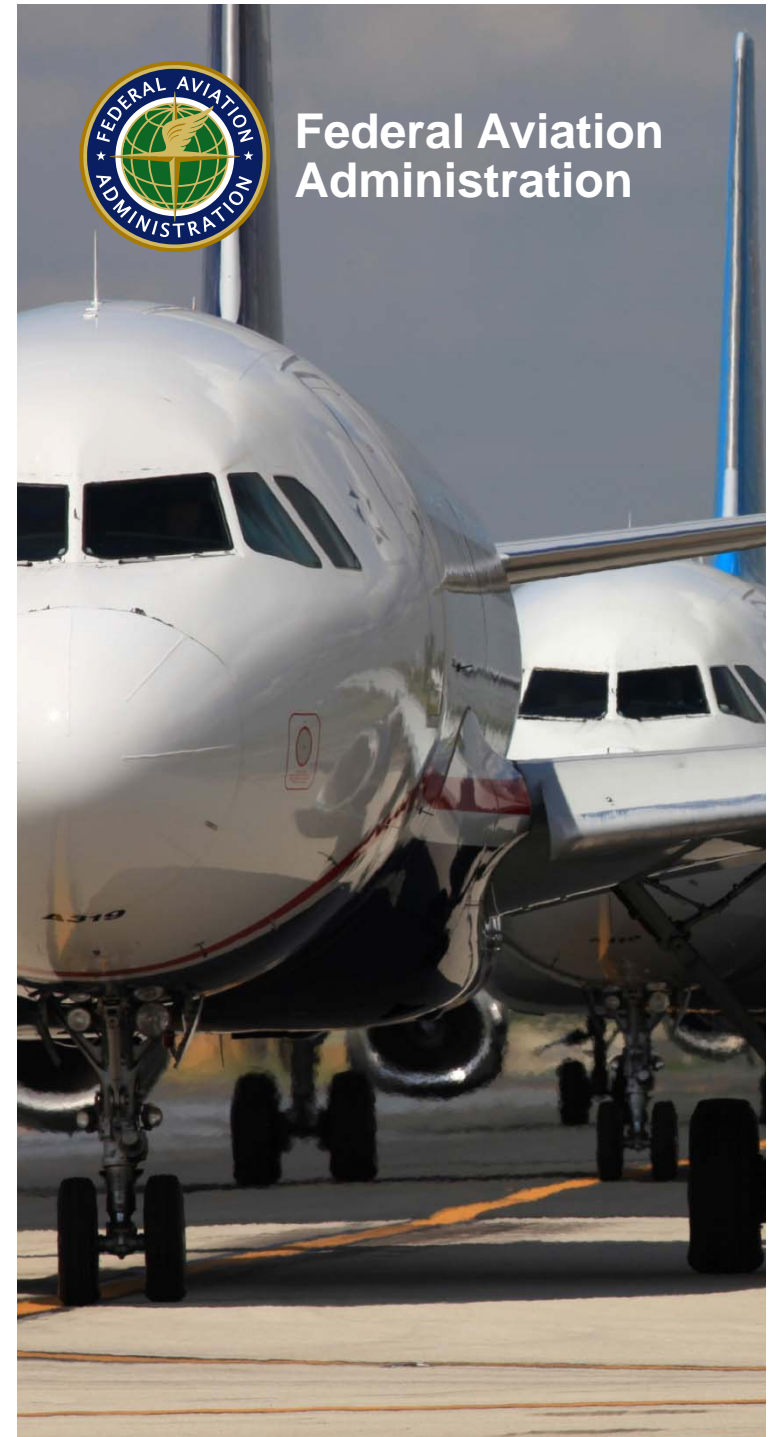


Safe Transport of PEDs in Transport Passenger Aircraft

Presented to: ICAO DGP
By: Fire Safety Branch
Federal Aviation Administration
Date: 10/20/2017



First and Foremost: Fire Prevention!

Aircraft are Designed, Certified, and Operated with the Philosophy of Preventing Accidents, which includes Preventing Any and All Fires from Occurring.



Cargo Fire Incidents (2002-2012)

Passenger Aircraft – Class “C” Compartments

- **N Registered Aircraft**
- **3 Incidents (2 Fires)**
 - Hair spray released in compartment
 - Overheating electronic unit that was on.
 - Flashlight that was on and overheated



Why are PEDs with Lithium Batteries an Added Risk?

- **Lithium batteries are both an ignition source and a fuel.**
- **Lithium batteries have been a fire source in the cabin.**
- **Lithium battery fires may reduce the effectiveness of the fire suppression system.**
 - They produce hydrogen gas when in thermal runaway.
 - Thermal runaway can propagate from cell to cell unless cooled.



Tablets in a Galley Cart



Lithium Ion Batteries in a Cargo Container



Federal Aviation
Administration

Why do cells go into thermal runaway and start fires?

- Over charged
- Discharge too fast
- Overheating
- Internal short (defective cell)
- Damage (punctured, dropped, etc.)



Fire Suppression System

- **Halon system is the second line of defense.**
 - Designed for fires likely to occur
- **Lithium batteries were not considered in design of system.**
- **Halon system may or may not be effective in controlling PED fires, i.e., the reliability of the system is negatively influenced by PED fires.**



Why Might Halon Not be Effective?

- **Thermal runaway can propagate from cell to cell, and Halon is not a good cooling agent.**
- **Cells in thermal runaway produce hydrogen, and the design concentrations of halon will not provide protection from a hydrogen explosion.**



Lithium Ion Battery Vent Gas Mixture

- Lithium batteries in thermal runaway produce flammable gasses and create significant hazards for aircraft.
- The three most prevalent gases are carbon dioxide (30.1 %), hydrogen (27.6 %), and carbon monoxide (22.9 %).
- Lithium ion battery vent gas mixture by percent concentration:

30.10% CO₂

27.60% H₂

22.90% CO

6.37% CH₄

4.48% C₃H₆

2.21% C₂H₄

1.57% C₄H₁₀

1.17% C₂H₆

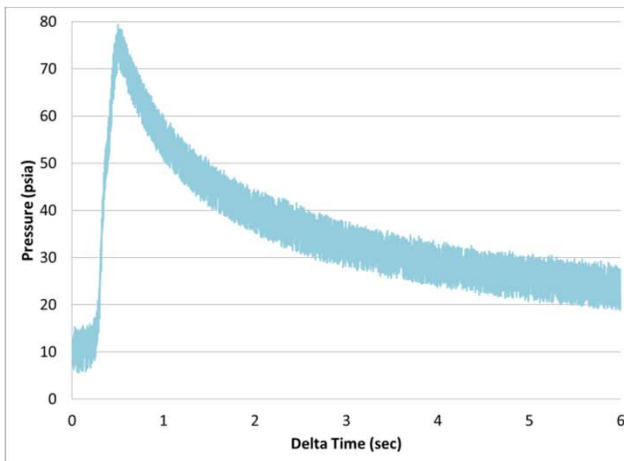
0.56% C₄H₈

0.27% C₃H₈

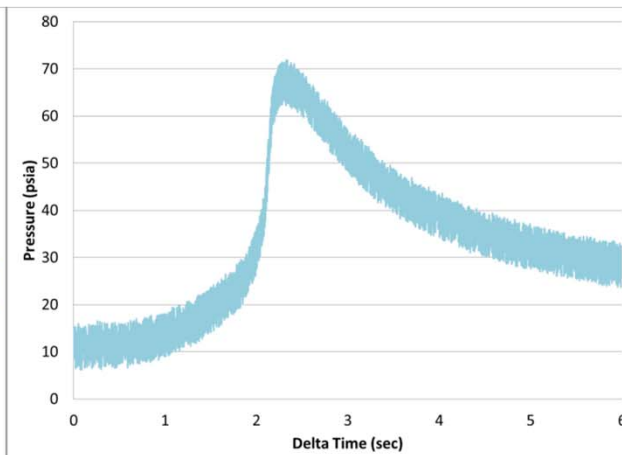


Results (Large Scale Tests)

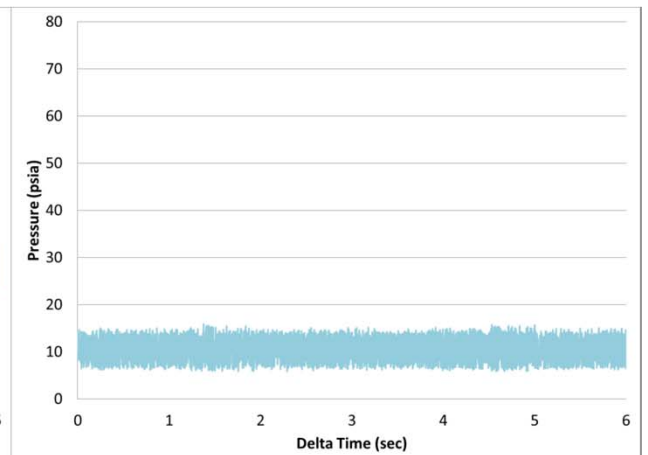
6 second time window



Test without
suppression



Test with
5.28% Halon



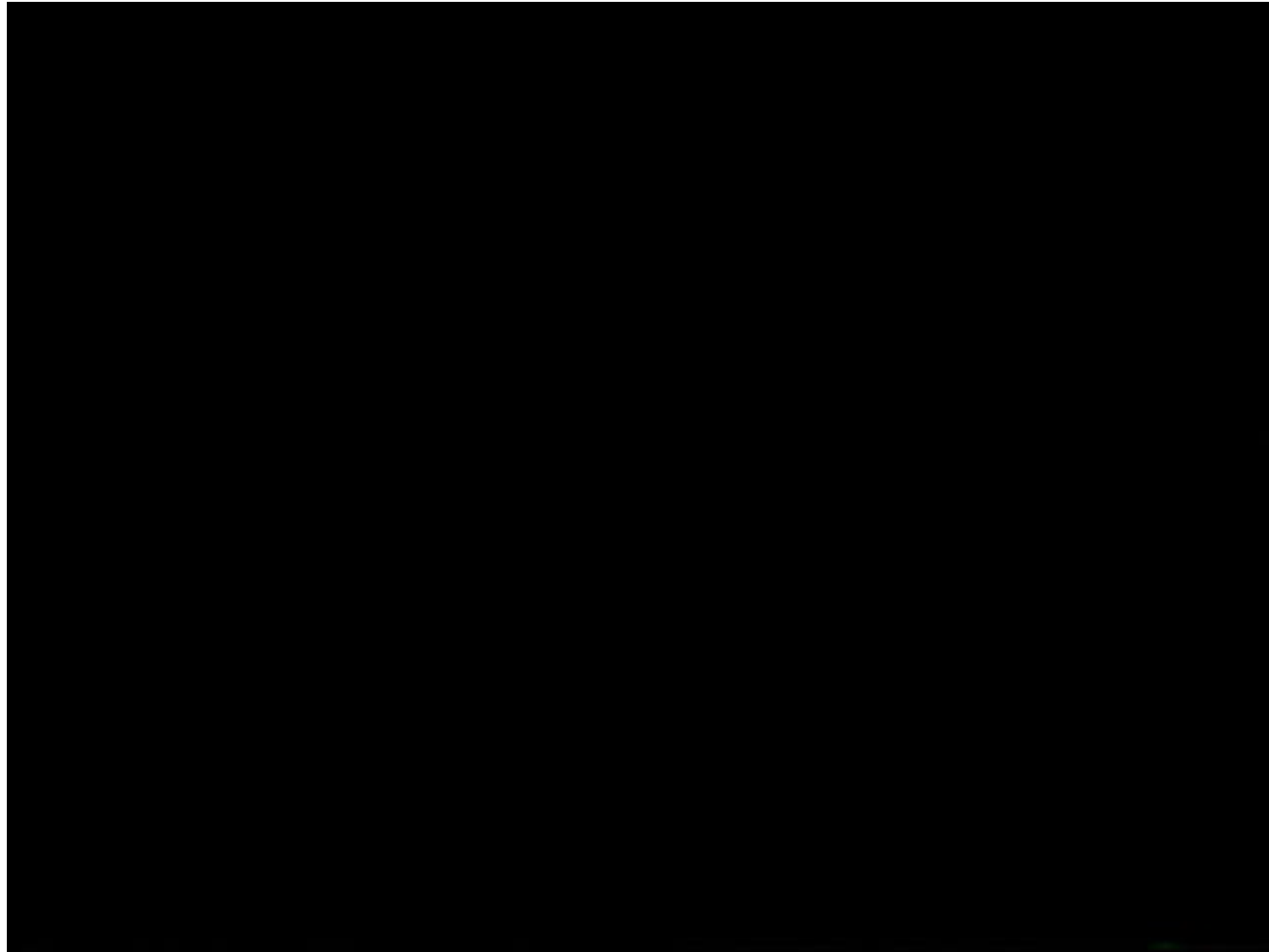
Test with
10.43% Halon

Effectiveness May Be Scenario Dependent

- **Fire buildup before detection**
- **Fire buildup before design concentration of agent penetrates container**
- **Fire involves other cargo**



Aerosol Can Explosion in a Class D Cargo Compartment



Why are Passenger PEDs a Greater Risk Than When Shipped by The Manufacturer?

- **From Manufacturer:**

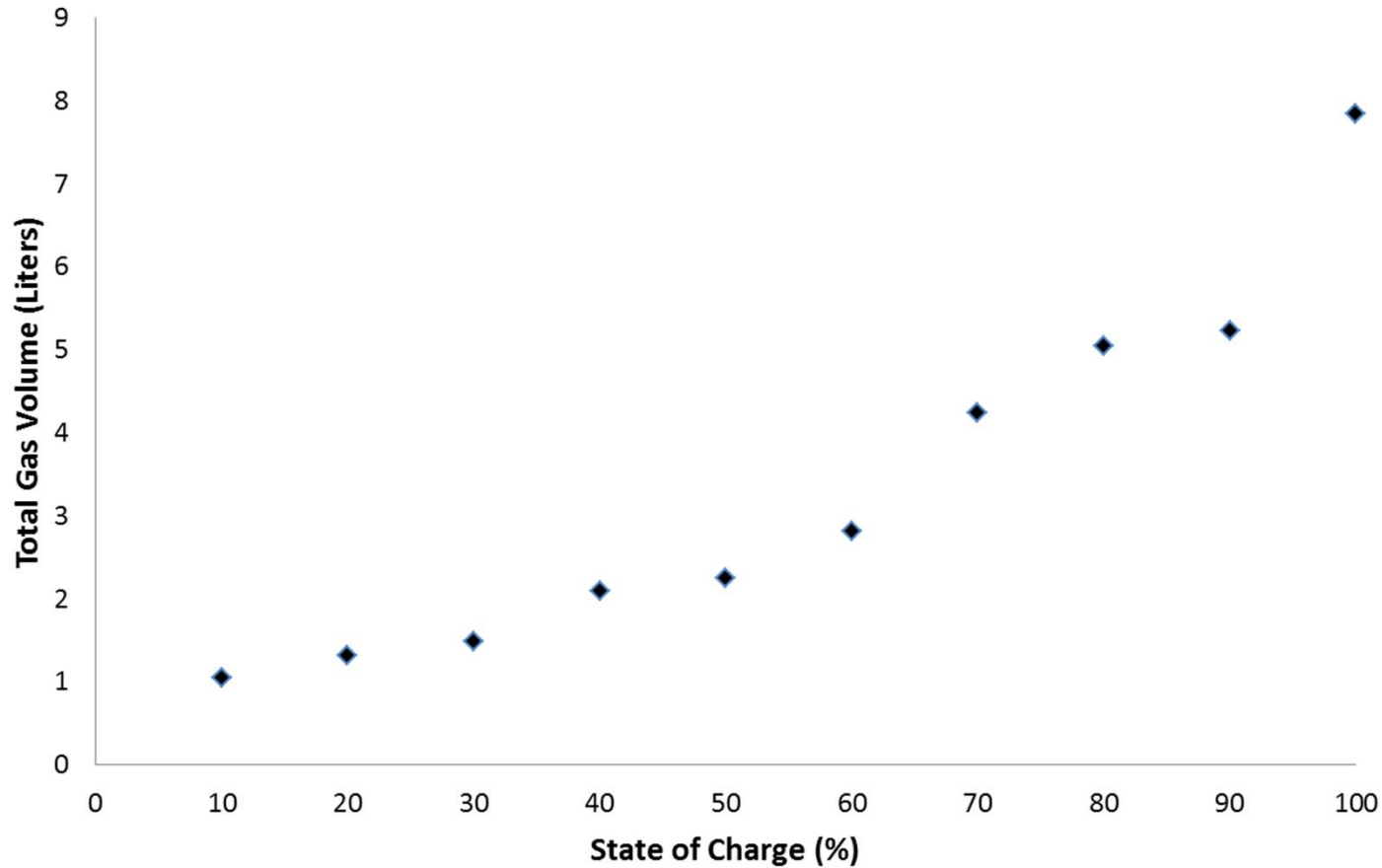
- Batteries not installed (unit always off).
- Batteries at reduced state of charge (most at 30%).
- Items are Marked and protected from damage.

- **From Passenger:**

- Batteries installed (unit might be in sleep mode).
- In most cases batteries are highly charged.
- Minimal protection from damage.
- Used (and potentially abused).



Gas Volume vs State of Charge



How to Mitigate or Reduce the Risk?

- **Prevention**

- Contain the event within one laptop package

- **Control**

- Control a PED fire such that the halon system can effectively suppress and contain the fire within the compartment.



Prevention

- **Operational Procedures:**
 - Laptops must be off. Much greater risk if on or in standby/sleep mode (can overheat the battery).
 - Packaged to prevent damage. (Any damage increases the risk of a fire).
 - Others as determined under **Control**.



Control:

- **Improve effectiveness of Halon system**
 - Optimal Placement of Laptops?
 - Best way to maximize halon at the laptop fire?
 - Quickest way to detect fire in container?

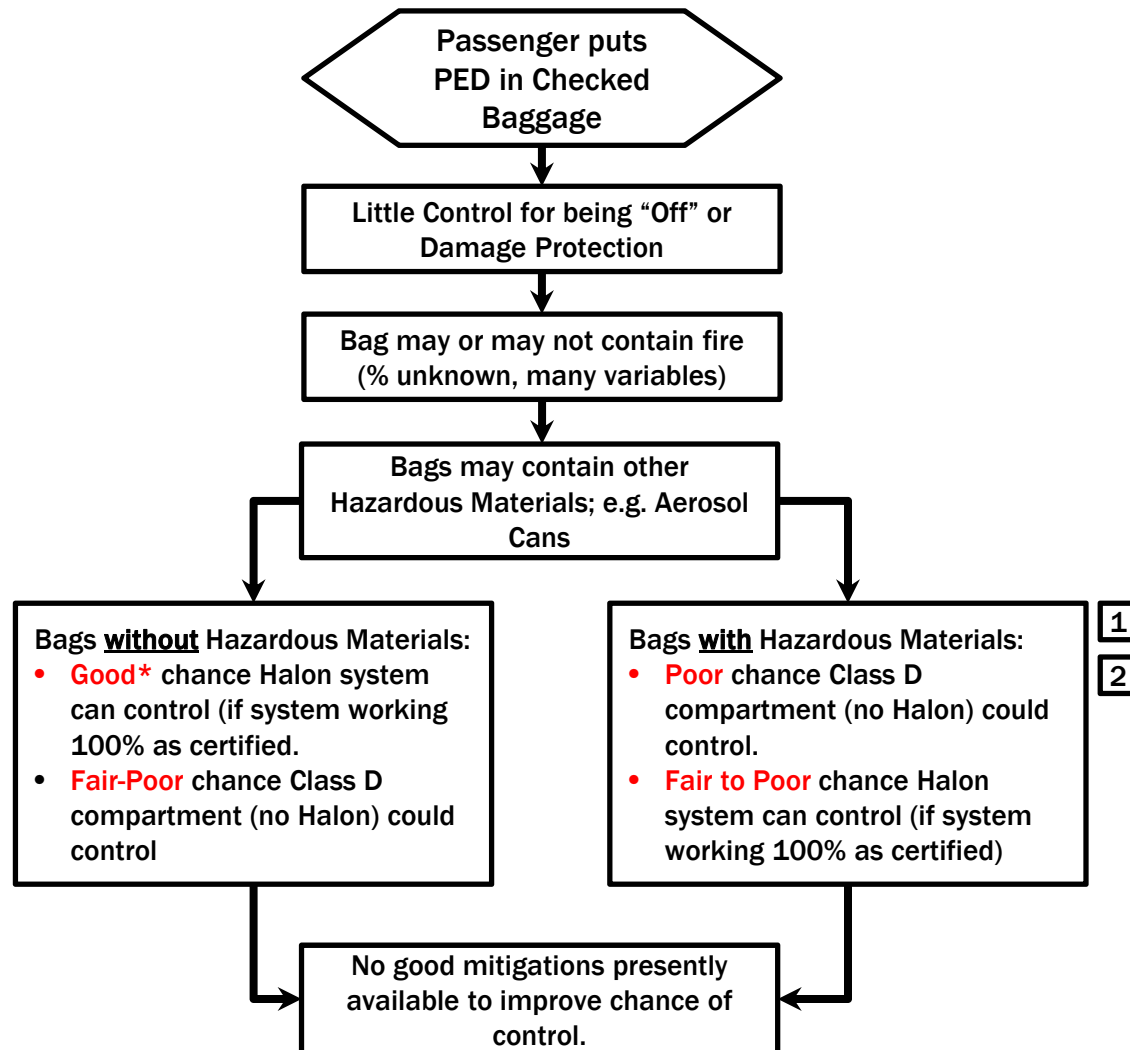


Summary:

- **Understand Risks**
- **Minimize Risks**
- **Determine if Minimized Risks are Acceptable**



PED's in Checked Bag



What Has Been Learned from Testing

Based on Laptops Powered by 18650 Cylindrical Cells

- **When packed in passenger bags:**
 - If a cell goes into thermal runaway the fire may penetrate the bag.
 - Depends on many factors, including, packing density, materials in bag, type of bag and available air in bag.
 - If other Haz. Mat.(such as aerosol cans) is in the bag an explosion can occur, not related to the gases from the battery.
 - An explosion can occur before Halon, in a Class C compartment, would be discharged and reach a 3% suppression concentration in a ULD.
 - A Class D compartment would provide no protection from an explosion



What Has Been Learned from Testing

Based on Laptops Powered by 18650 Cylindrical Cells

- **Large scale test demonstrated:**
 - About 2 minutes from agent discharge until 3% suppression concentration was reached within a ULD.
 - That time will depend on many factors, including agent discharge time, load factor of the compartment, leakage rate of the compartment and tightness of the ULD.
 - A Halon system in a class c compartment, if working as certified, can control the fire of boxes packed in a ULD.
 - A Halon system in a class c compartment, if working as certified, might not provide enough agent inside a ULD in time to suppress an explosion of an aerosol can caused by a laptop battery fire.



Pouch Cells Pose Same Risk



Laptop Luggage Test Overview

- **Luggage provided by TSL, fully loaded with various items of clothing, shoes, books and other personal items.**
- **Laptop fitted with heater and thermocouple to initiate thermal runaway.**
- **Laptop placed in bag with clothing items above and below the laptop.**
- **Thermal runaway is initiated and the bag monitored for smoke, open flames, and temperature.**



Test 1: Soft Sided Luggage

- Luggage was opened, some contents removed and the laptop was placed inside.
- The remaining contents were replaced on top of the laptop.



Test 1 Results

- The laptop was put into thermal runaway.
- Smoke was observed escaping the bag
- No open flames were observed.
- Some charring of bag contents was found post test.



Test 2: Hard Sided Luggage

- Luggage prepared in same manner as test 1.
- Large amounts of smoke were observed.
- No open flames were observed



Test 2: Results

- **Considerable charring of contents.**
- **Some damage to inside of bag .**
- **Small penetration on the underside of the bag**



Test 3: Soft Sided Bag

- Smoke observed.
- No open flames were observed.
- Charring of contents
- Damage to inner liner

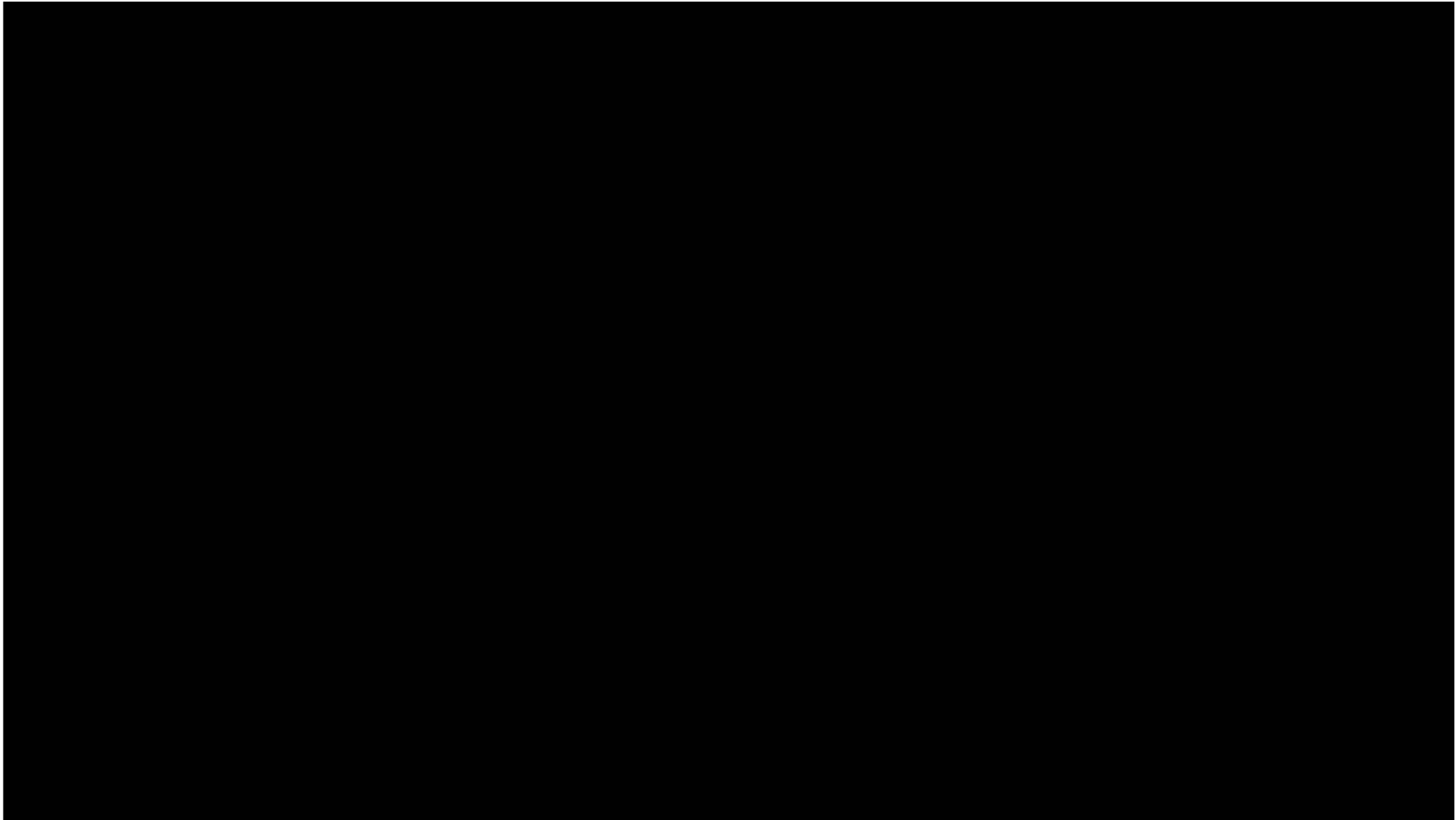


Test 4: Soft Sided Luggage Results

- Smoke observed at thermal runaway.
- Bag was breached and open flames were observed.
- Hot vigorous fire developed.
- Bag completely destroyed.



Test 4: Soft Sided Luggage Results



Test 5: Small Duffel Bag Results

- Smoke observed.
- No open flames were observed.
- Charring of contents
- Some slight burning/charring through to exterior of bag



Observations

- **Tightly packed suitcases seem to contain the laptop fire better than loosely packed suitcases**
- **Laptop fire will spread to the suitcase contents if there is a sufficient air source, such as a hole in the bag.**
- **The type of bag contents has an effect on flammability.**
- **There does not appear to be a difference between soft sided and plastic hard sided bags in terms of containing a laptop fire.**
- **Metal sided suitcases were not available for testing.**



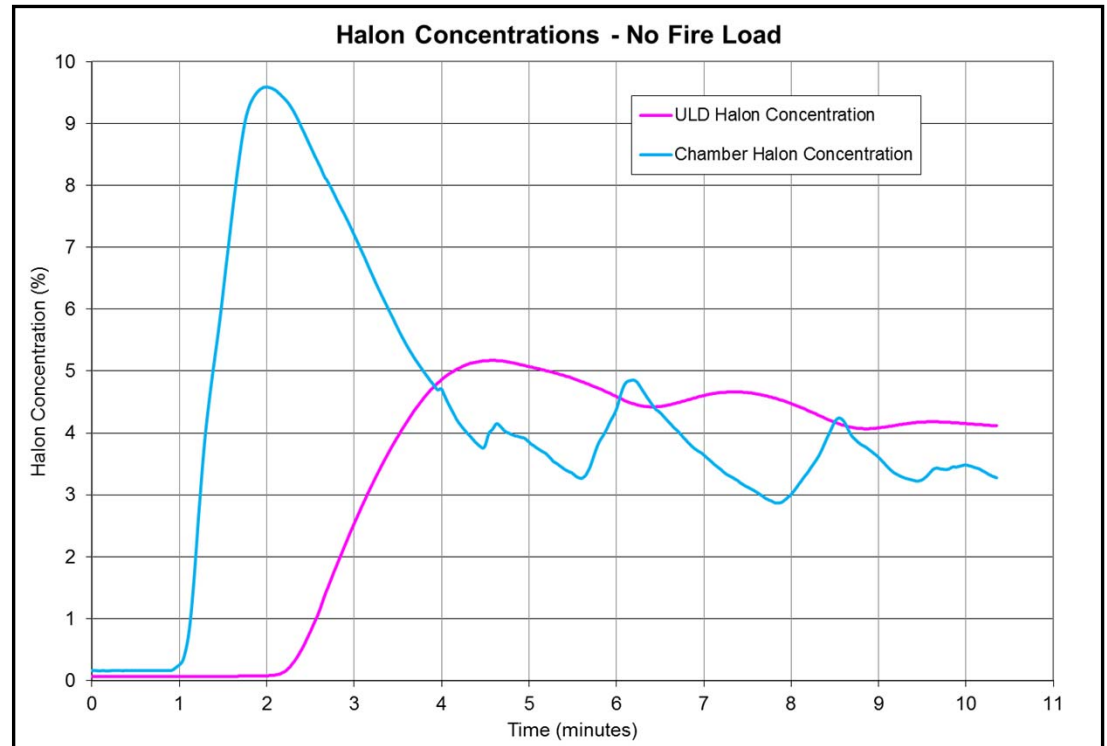
Test Setup: Simulate Class C Cargo Compartment

- 381 cubic foot test chamber.
- 32 cubic foot ULD
- Halon system installed with 5% initial concentration and 3% maintained concentration
- Forced leakage rate 10 cubic feet/minute
- 60-70 % cargo loading by volume



Halon Concentration Test

- Halon was measured in two locations:
 - Outside the ULD near mid chamber height
 - Inside the ULD near mid chamber height
- 7.5 lb Halon discharge yields 5% in the chamber when empty
- Peak concentration is higher due to chamber loading.
- Additional Halon added as concentration depleted to maintain 3-4%



Laptop Fire Test with Halon

- **36 laptops, Emirates style packaging**
 - Charged to 100%
 - Packed in bubble wrap inside Emirates style box
- **Laptop in position 9 fitted with a heater to induce thermal runaway**
- **Halon discharged when visible smoke is observed**
- **3% Halon concentration maintained for duration of test.**



Results: Laptop Fire with Halon

- Heater was energized at time 0.
- Thermal runaway was detected at 8:21
- Smoke was observed at 8:21
- Halon was discharged at 9:50
- Halon maintenance began at 15:30
- Peak Halon concentration in the chamber was 9.25%
- Peak Halon concentration in the ULD was 7.65%
- Test terminated after 60 minutes



Results: Laptop Fire with Halon

- Soot visible on the boxes above the laptop in thermal runaway in box 9.
- Bottom of box 9 burned through



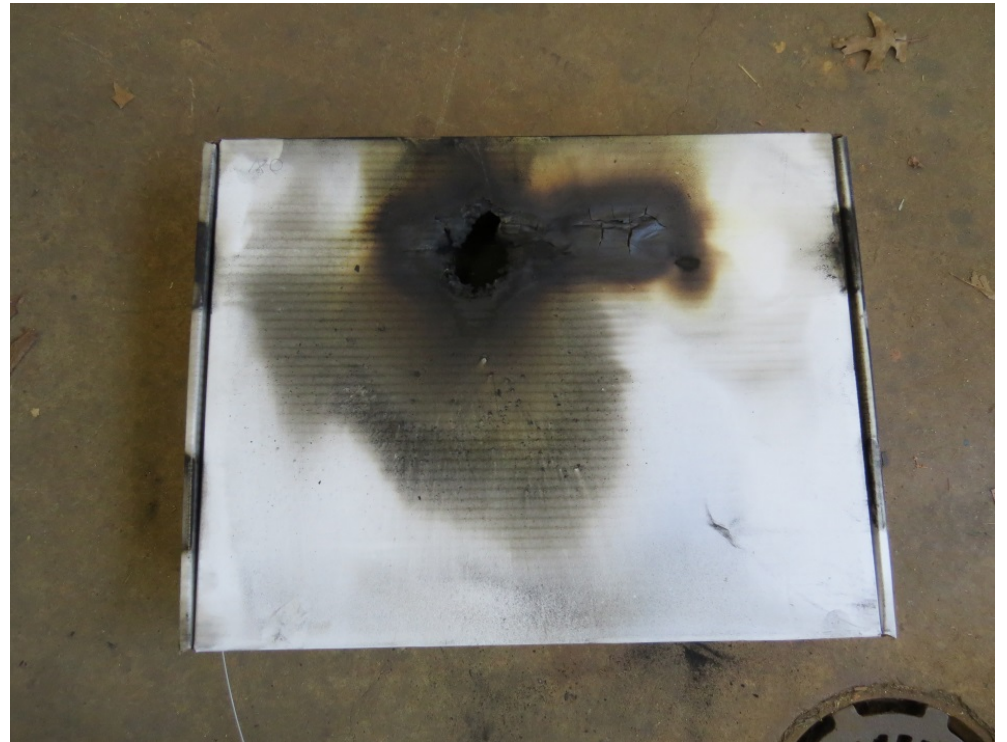
Results: Laptop Fire with Halon

- **Laptop in thermal runaway**
 - Bubble wrap on top charred
 - Bubble wrap on bottom consumed
 - All cells in battery pack went into thermal runaway



Results: Laptop fire with Halon

- Lid of box 10, located directly below box 9, was penetrated.



Results: Laptop Fire with Halon

- **Laptop in box 10**
 - Bubble wrap melted.
 - Laptop was charred.
 - Temperature data for this laptop indicated the interior of the laptop never exceeded 100°F.



Observations: Laptop Test

- Halon was able to penetrate the simulated ULD and achieved a sufficient concentration to suppress the fire.
- The laptop in thermal runaway generated enough heat to both char and penetrate the bottom of the box and the top of the box below it.
- There was no propagation of thermal runaway to adjacent laptops



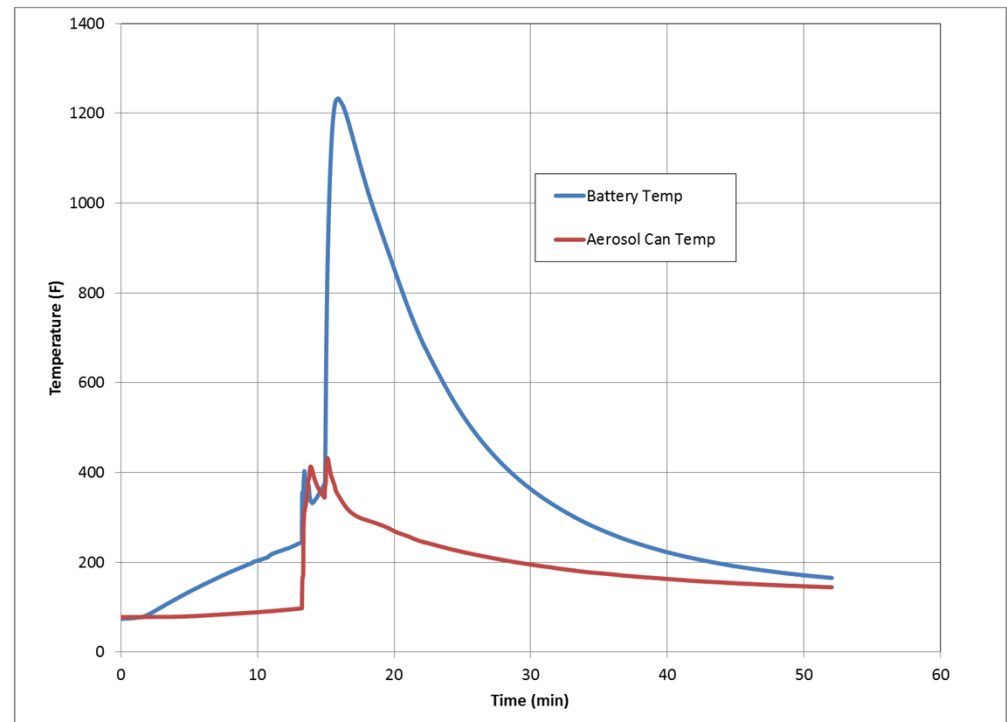
Laptop/Aerosol Can Test 1

- Laptop prepared as before.
- 12 ounce aerosol can of hairspray strapped to laptop battery to ensure it stays in close proximity
- Laptop/can placed in hard sided suitcase



Laptop/Aerosol Can Test 1 Results

- Thermal runaway of battery resulted in some smoking.
- Temp reading of aerosol can appears to have been affected by the TR event.
- This temperature remained well above 200F for ~15 minutes
- No resulting explosion of can



Laptop/Aerosol Can Test 2

- Laptop prepared as before.
- 8 ounce aerosol can of dry shampoo strapped to laptop battery to ensure it stays in close proximity
- Laptop/can placed in soft sided suitcase

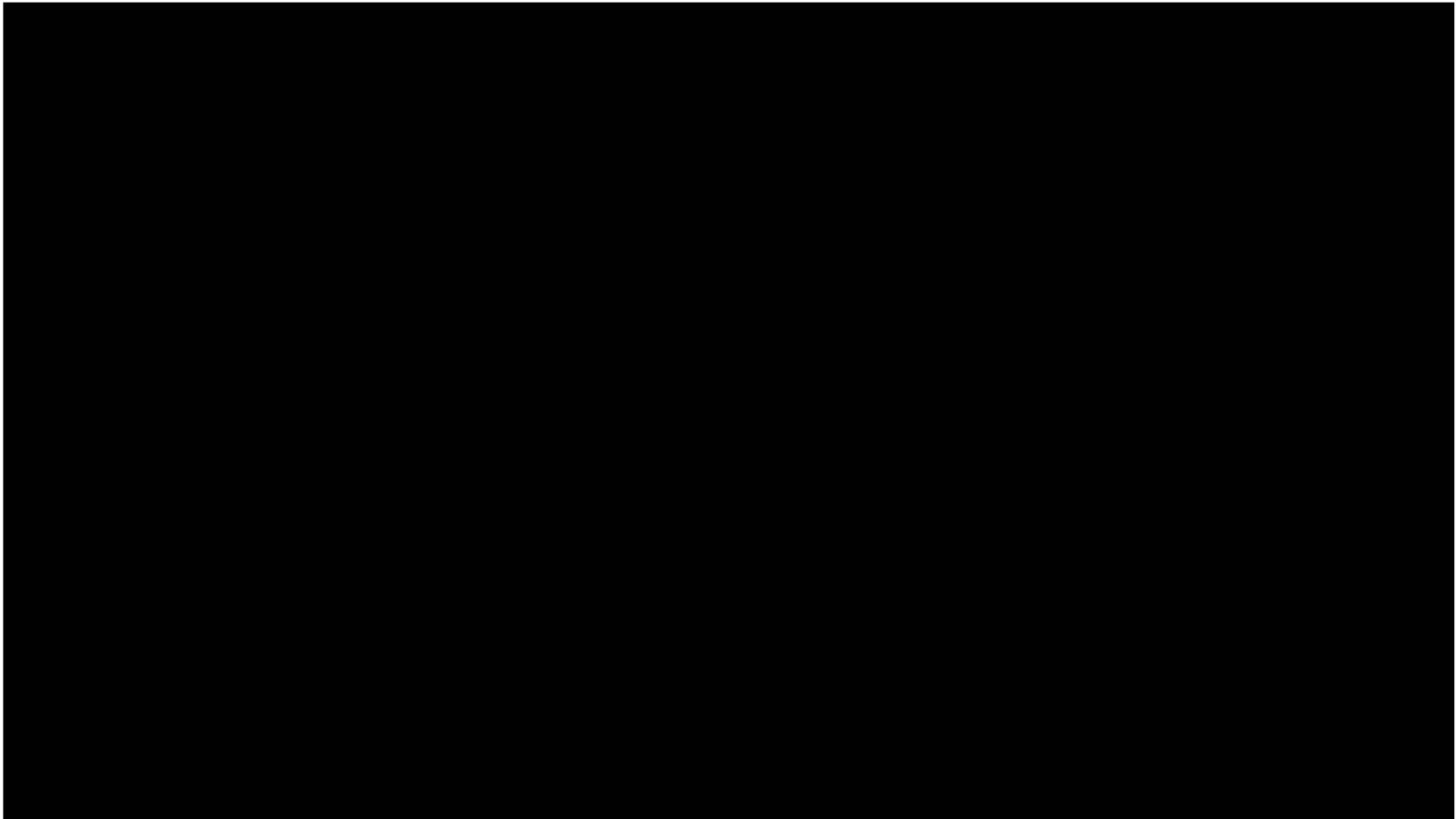


Laptop/Aerosol Can Test 2 Results

- Fire observed almost immediately after first thermal runaway event.
- Fire rapidly grew and within 40 second can exploded
- Fire continued to rapidly consume bag/contents



Laptop/Aerosol Can Test



Observations

- If an aerosol can is packed in a suitcase and a thermal runaway event occurs, there is the potential for an aerosol can explosion.

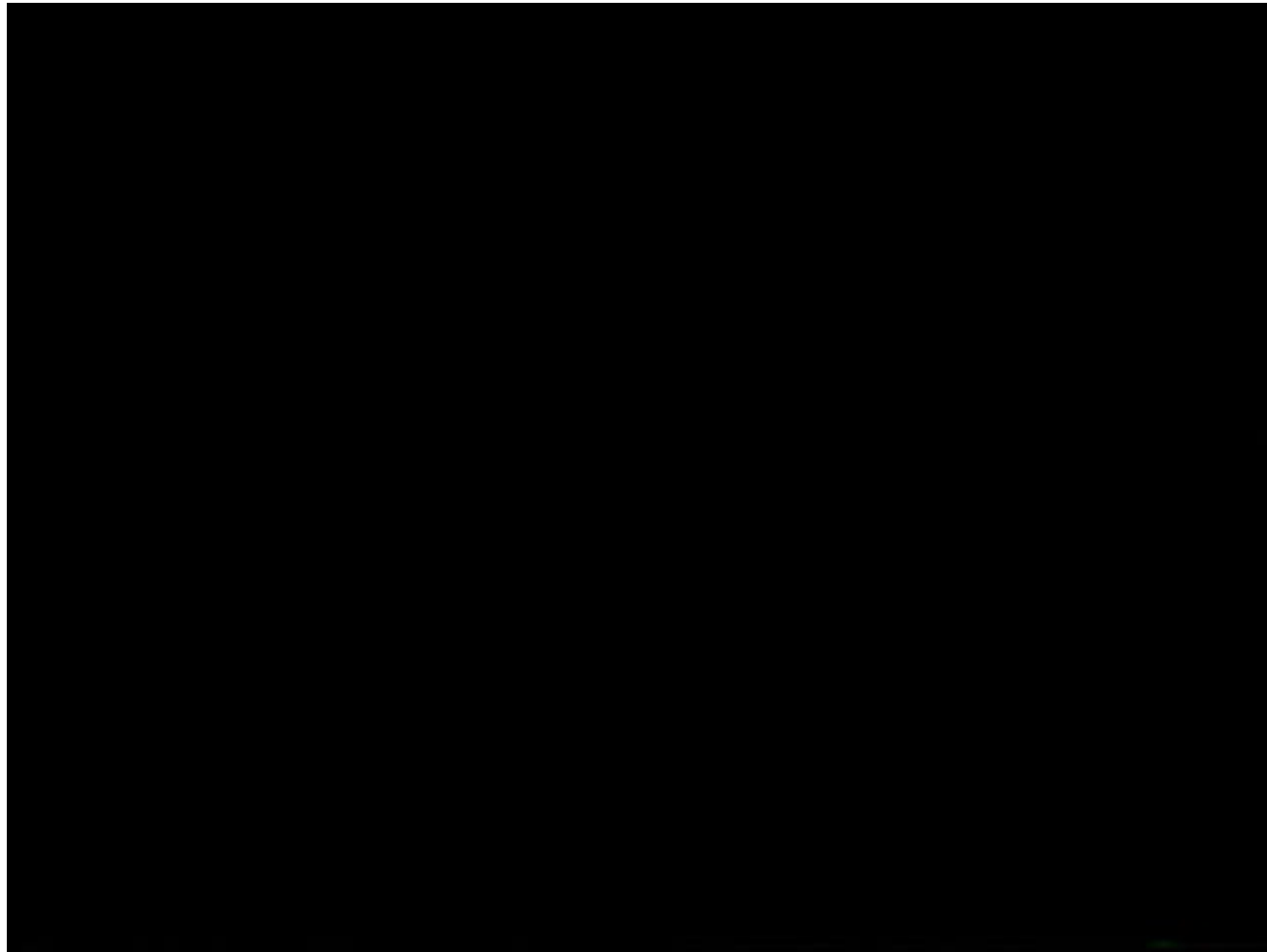


Effectiveness May Be Scenario Dependent

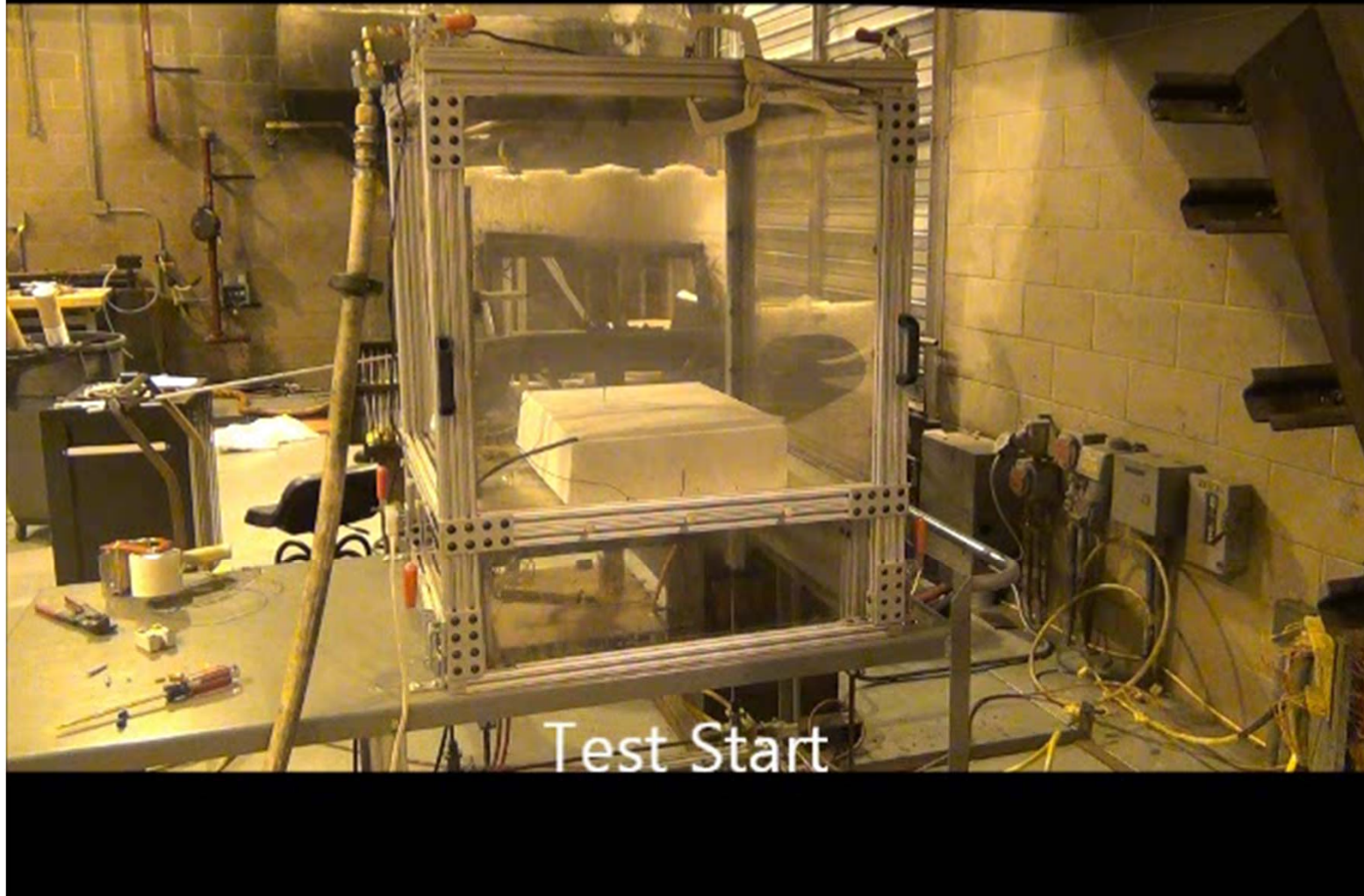
- **Fire buildup before detection**
- **Fire buildup before design concentration of agent penetrates container**
- **Fire involves other cargo**



Aerosol Can in Class D Compartment



Laptop in Box Lined With Fire Barrier Tested to G27 Proposed Standard



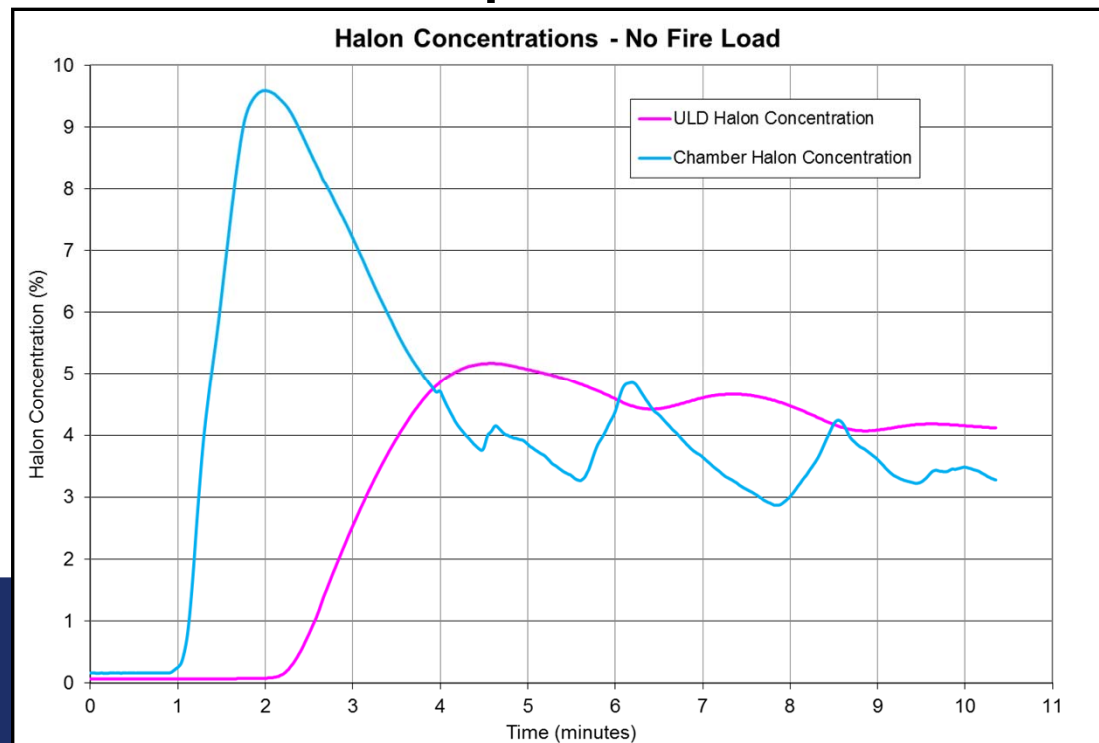
Test Setup: Simulate class C cargo compartment

- 381 cubic ft test chamber
- 32 cubic ft. ULD
- Halon system installed. 5% initial concentration, 3% maintained
- Forced leakage rate 10 cubic feet/minute
- 60-70 % cargo loading by volume



Halon concentration test

- Halon was measured in two locations:
 - Outside the ULD near mid chamber height
 - Inside the ULD near mid chamber height
- 7.5 lb Halon discharge yields 5% in the chamber when empty
- Peak concentration is higher due to chamber loading.
- Additional Halon added as concentration depleted to maintain 3-4%



Luggage fire test with Halon

- 5 bags of luggage, filled with misc. clothing and personal effects.
- Ignition source in cardboard box filled with shredded paper placed in center position
- Halon discharged when visible smoke is observed
- 3% Halon concentration maintained for duration of test.



Results: Luggage fire test with Halon

- Heavy smoke observed shortly after halon discharged – did not clear until after ~1 hour
- Cardboard box not fully consumed
- Soot/smoke particles observable throughout pressure vessel post-test



Results: Luggage fire test with Halon

- Peak temperatures inside the 5 pieces of luggage ranged from 120 - 150°F
- 2 of the 5 bags had minor charring.



Observations- Luggage test

- Halon was able to penetrate the simulated ULD and achieve a sufficient concentration to suppress the fire.
- Fire did not propagate from cardboard box to the pieces of luggage, however some charring did occur.
- Heavy smoke throughout pressure vessel for the full hour duration of test due to likely smoldering fire within cardboard box.
- Halon performed as expected

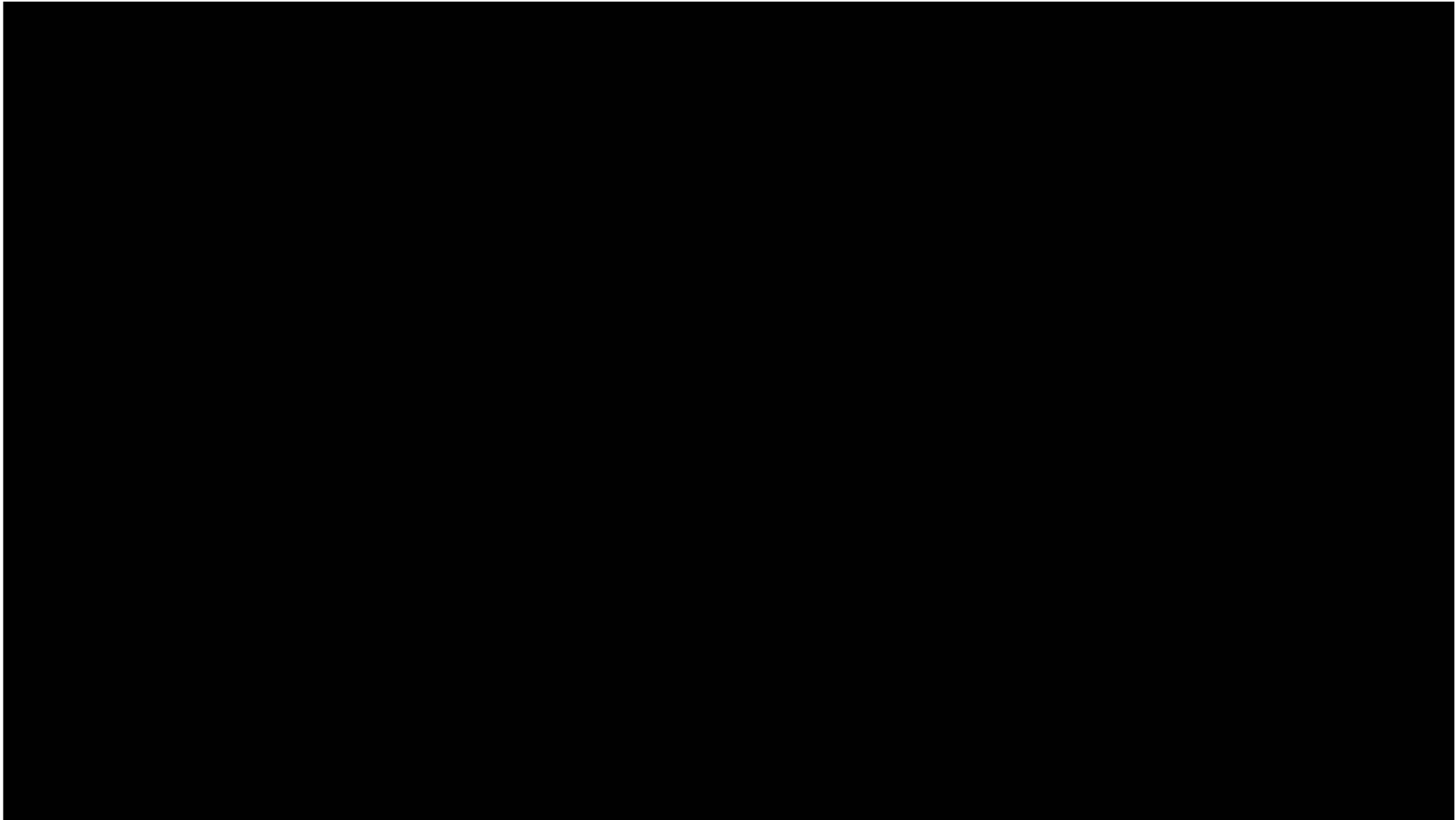


6-22-17 Luggage Test #1 - Results

- Fire visible approximately 20 s after initial TR event
- Approximately 8.5 minutes after fire initiated, aerosol can exploded.



6-22-17 Luggage Test #1 - Results



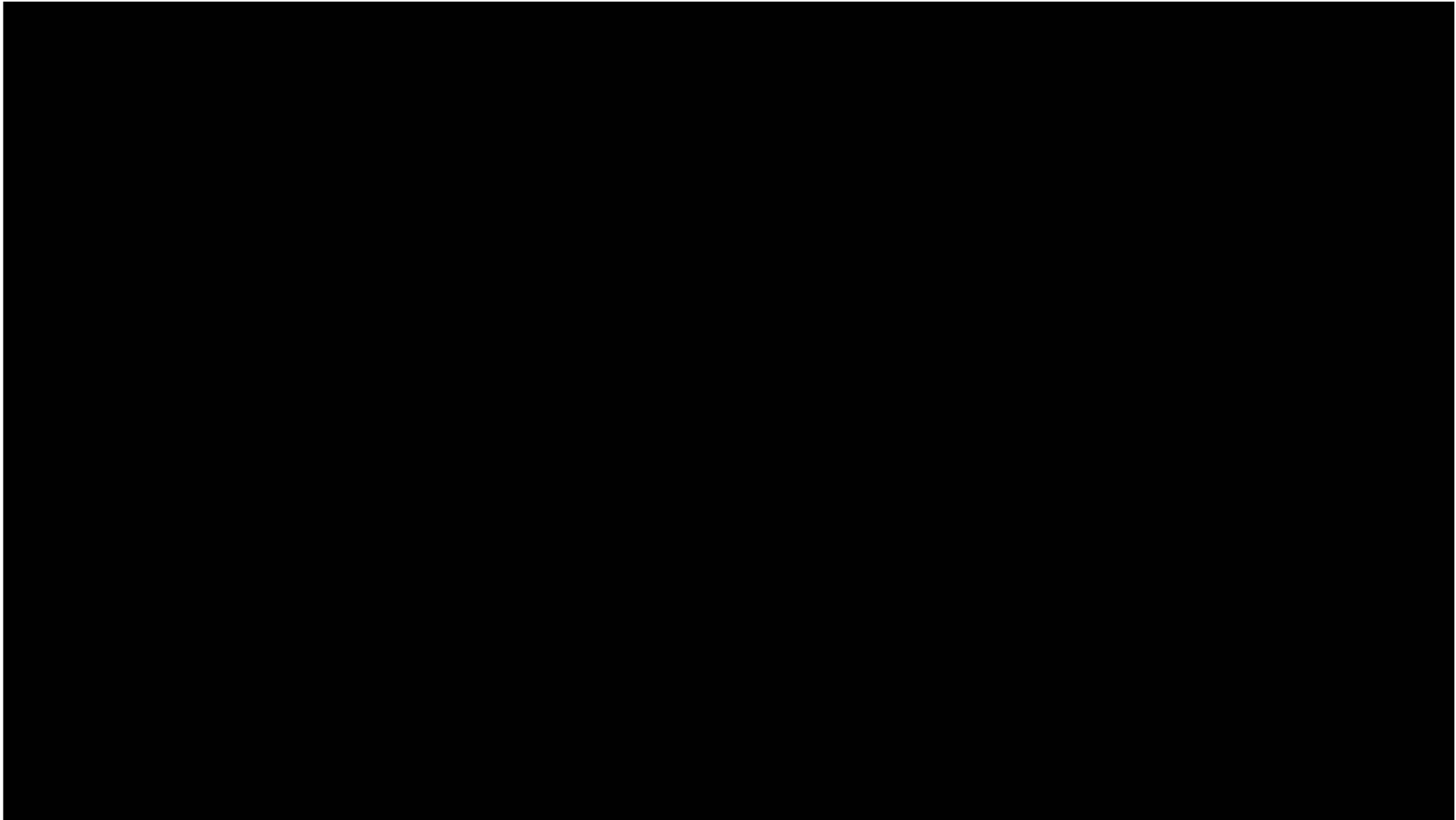
Federal Aviation
Administration

6-22-17 Luggage Test #2 - Results

- 1st evidence of TR observed ~18.5 mins after heater activation.
- Heavy smoke coming from bag following TR.
- No evidence of flames.
- No involvement of aerosol can.
- Burning/charring of contents evidenced post-test.



6-22-17 Luggage Test #2 - Results



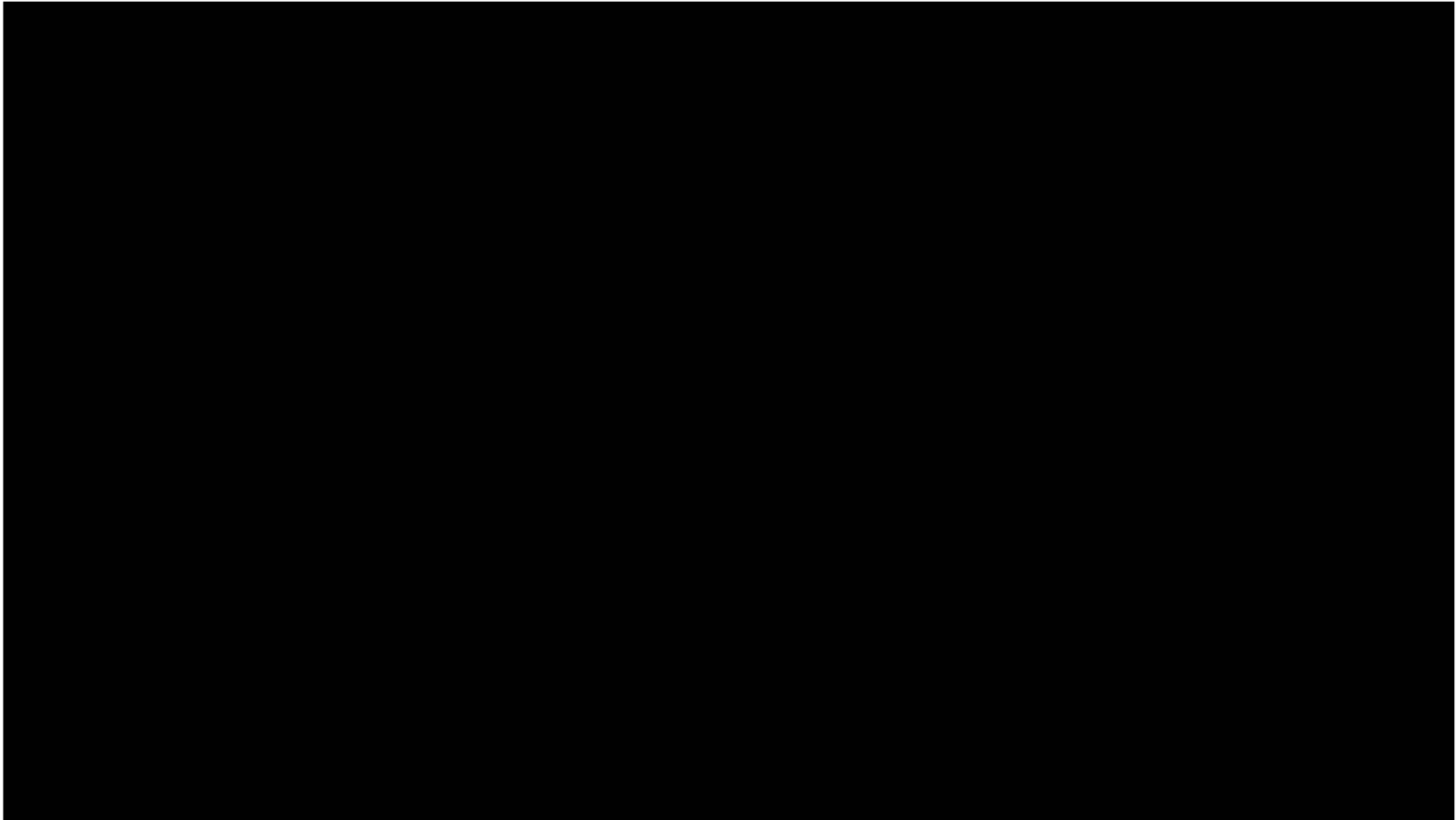
Federal Aviation
Administration

6-22-17 Luggage Test #3 - Results

- Initial TR event occurs with release of smoke/flame
- Fire progresses and within 10 seconds, aerosol can explodes.



6-22-17 Luggage Test #3 - Results



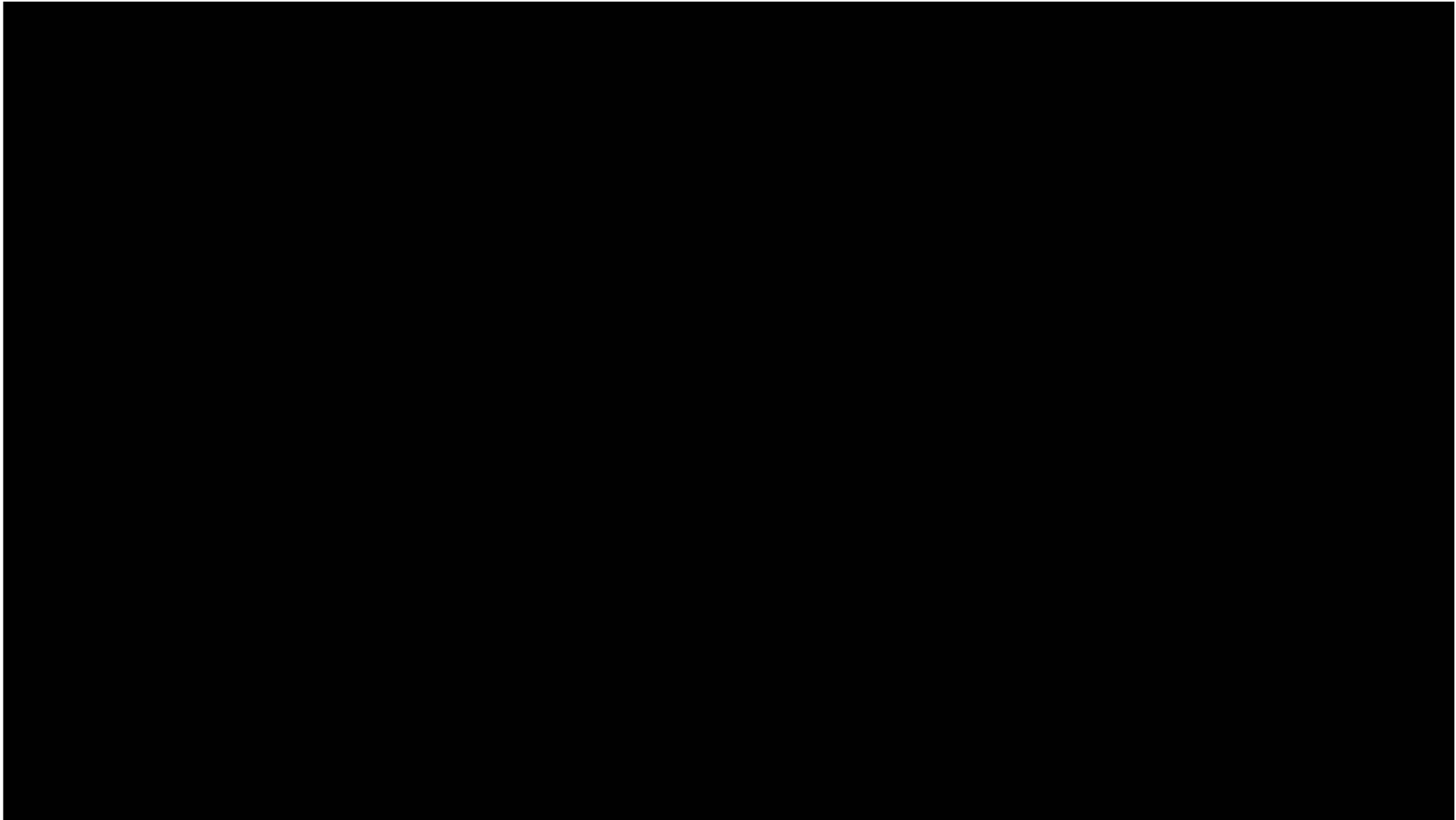
Federal Aviation
Administration

6-22-17 Luggage Test #4 - Results

- ~10 mins after start of test, first TR event occurs
- Fire immediately visible
- Burning continues for 5-6 minutes followed by violent aerosol can explosion



6-22-17 Luggage Test #4 - Results



Federal Aviation
Administration

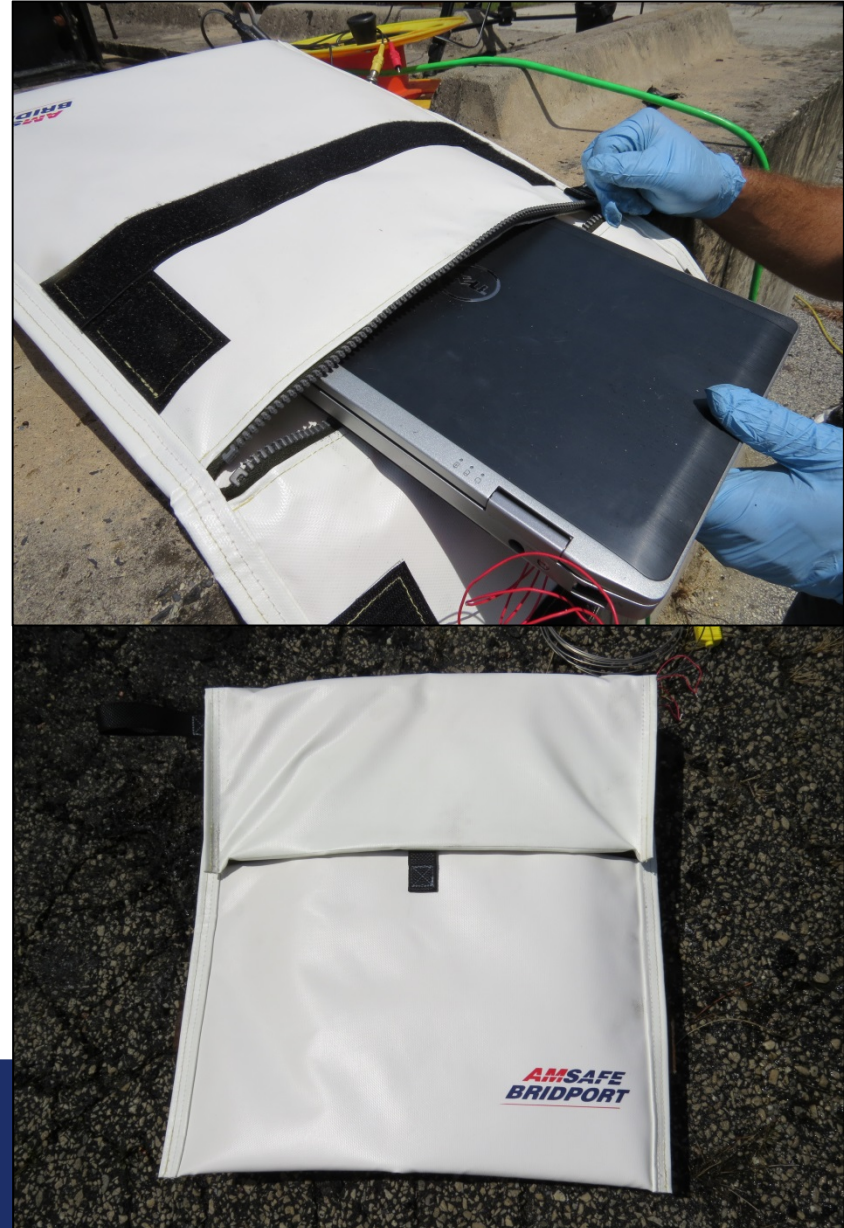
Luggage Test Observations

- **Tightly packed suitcases seem to contain the laptop fire better than loosely packed suitcases**
- **Laptop fire will spread to the suitcase contents if there is a sufficient air source, such as a hole in the bag.**
- **The type of bag contents has an effect on flammability.**
- **The presence of flammable toiletries increases the potential for fire.**
- **There does not appear to be a difference between soft sided and plastic hard sided bags in terms of containing a laptop fire.**
- **Metal sided suitcases were not available for testing.**
- **If an aerosol can is packed in suitcase and a thermal runaway event occurs, there is the potential for an aerosol can explosion.**



6-22-17 AMSafe Pouch Test

- Laptop instrumented as previous tests
- Placed in pouch
- Pouch zippered and sealed.
- Additional T/C positioned inside flap of the pouch

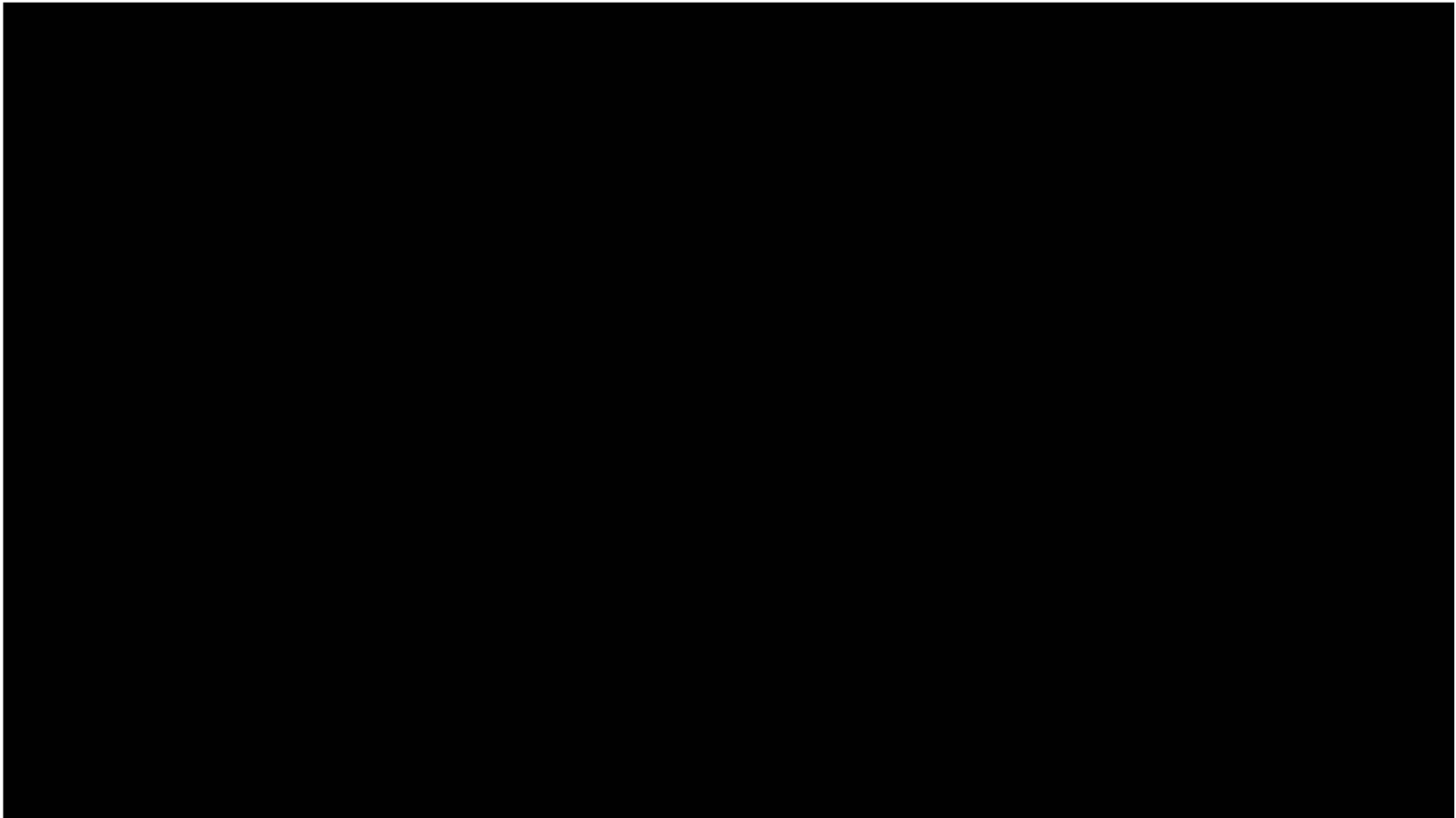


6-22-17 AMSafe Pouch Test - Results

- Significant smoke seen emanating from pouch as TR events occurred
- Pouch expanded during TR events due to pressure build up
- Flames escaped from pouch ~30 sec after initial TR event
- Post-test evaluation showed a rupture had occurred in back surface of bag.



6-22-17 AMSafe Pouch Test - Results



Federal Aviation
Administration

